

# Safety Data Sheet

# Ammonium Hydroxide 28%

## **SECTION 1: Identification**

#### 1.1 GHS Product identifier

Product name Ammonium Hydroxide 28%

Substance name Ammonia, aqueous solution

EC no. 215-647-6 CAS no. 1336-21-6 Index no. 007-001-01-2

#### 1.2 Other means of identification

Aqueous ammonia, Ammonia solution

### 1.3 Recommended use of the chemical and restrictions on use

For laboratory and manufacturing use only.

### 1.4 Supplier's details

Name High Purity Products
Address 14546 N. Lombard Street
Portland OR 97203
United States of America

Telephone 503-227-1616

email help.desk@highpp.com

### 1.5 Emergency phone number

CHEMTREC: 1-800-424-9300

### **SECTION 2: Hazard identification**

#### 2.1 Classification of the substance or mixture

GHS classification in accordance with: OSHA (29 CFR 1910.1200)

- Acute toxicity, oral, Cat. 4
- Acute toxicity, inhalation, Cat. 4
- Skin corrosion/irritation, Cat. 1B

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- Serious eye damage/irritation, Cat. 1

#### 2.2 GHS label elements, including precautionary statements

#### **Pictogram**



Signal word Danger

Hazard statement(s)

H302 Harmful if swallowed H332 Harmful if inhaled

H314 Causes severe skin burns and eye damage

H318 Causes serious eye damage

**Precautionary statement(s)** 

P234 Keep only in original container.

P260 Do not breathe vapors.

P262 Do not get in eyes, on skin, or on clothing.
P264 Wash hands thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P284 In case of inadequate ventilation wear respiratory protection.
P301+P310 IF SWALLOWED: Immediately call POISON CENTER

P302+P361+P354 IF ON SKIN: Take off Immediately all contaminated clothing. Immediately rinse

with water for 15 minutes.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P305+P351+P338 IF IN EYES: Rinse cautiously with water for 15 minutes. Remove contact lenses if

present and easy to do. Continue rinsing.

P361+P364 Take off immediately all contaminated clothing and wash it before reuse.

P390 Absorb spillage to prevent material damage.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

### **SECTION 3: Composition/information on ingredients**

#### 3.1 Mixture

Components	CAS#	Percent (weight)
Ammonium Hydroxide	1336-21-6	27-29%
Water	7732-18-5	71-73%

### **SECTION 4: First-aid measures**

### 4.1 Description of necessary first-aid measures

If inhaled Remove the victim into fresh air. Respiratory problems: consult a doctor/medical

service.

In case of skin contact Wash immediately with lots of water (15 minutes)/shower. Do not apply (chemical)

neutralizing agents. Remove clothing while washing. Do not remove clothing if it

sticks to the skin. Cover

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wounds with sterile bandage. Consult a doctor/medical service. If burned surface >

10%: take victim to hospital.

In case of eye contact Rinse immediately with plenty of water for 15 minutes. Cover eyes aseptically. Do

not apply neutralizing agents. Take victim to an ophthalmologist.

If swallowed Do NOT induce vomiting. Never give anything by mouth to an unconscious person.

Rinse mouth with water. Consult a physician.

#### 4.2 Most important symptoms/effects, acute and delayed

Causes burns by all exposure routes. Ingestion causes severe swelling, severe damage to the delicate tissue and danger of perforation: Product is a corrosive material. Use of gastric lavage or emesis is contraindicated. Possible perforation of stomach or esophagus should be investigated.

### **SECTION 5: Fire-fighting measures**

#### 5.1 Suitable extinguishing media

EXTINGUISHING MEDIA FOR SURROUNDING FIRES: All extinguishing media allowed

### 5.2 Specific hazards arising from the chemical

Nitrogen oxides (NOx)

Not combustible.

Ammonia solution itself is not flammable, but can form an ignitable ammonia/air-mixture by outgassing.

Ambient fire may liberate hazardous vapors.

Fire may cause evolution of: nitrogen oxides

#### 5.3 Special protective actions for fire-fighters

Stay in danger area only with self-contained breathing apparatus. Prevent skin contact by keeping a safe distance or by wearing suitable protective clothing.

### **SECTION 6: Accidental release measures**

### 6.1 Personal precautions, protective equipment and emergency procedures

Wear respiratory protection. Avoid breathing vapors, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas.

#### **6.2** Environmental precautions

Should not be released into the environment.

#### 6.3 Methods and materials for containment and cleaning up

Soak up with inert absorbent material and dispose of as hazardous waste. Keep in suitable, closed containers for disposal.

# **SECTION 7: Handling and storage**

## 7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid inhalation of vapor or mist.

#### 7.2 Conditions for safe storage, including any incompatibilities

No metal or light-weight-metal containers.

Keep container tightly closed in a dry and well-ventilated place. Containers which are opened must be carefully resealed and kept upright to prevent leakage.

### **SECTION 8: Exposure controls/personal protection**

### 8.1 Control parameters

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#### Ammonia (CAS # 7664-41-7)

NIOSH REL: TWA 25 ppm (18 mg/m3)

### 8.2 Appropriate engineering controls

Avoid contact with skin, eyes and clothing. Wash hands before breaks and immediately after handling the product.

#### 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### **Pictograms**









### Eye/face protection

Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US).

### Skin protection

Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

#### **Body protection**

Complete suit protecting against chemicals, Flame retardant antistatic protective clothing., The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace.

### Respiratory protection

Gas mask with filter type K. High vapor/gas concentration: self-contained respirator.

### SECTION 9: Physical and chemical properties and safety characteristics

Physical state Liquid

Appearance Clear, Colorless

Color Clear

Odor Strong, ammonia
Odor threshold 5 - 50 ppm

pH 13.8

Melting point/freezing point -69.2 °C (-92.6 °F)
Boiling point or initial boiling point and boiling range 38-100°C (100-212°F)

Flash point Standard Solling Family

Flash point Not Flammable

Evaporation rate

Flammability

Not Flammable

Not Flammable

Vapor pressure 476.29 mmHg at 20°C (68°F)

Relative vapor density

0.6

Density and/or relative density

0.90

Solubility Water: Complete

Partition coefficient n-octanol/water (log value)

No Data Available

Auto-ignition temperature

No Data Available
651°C

Decomposition temperature No Data Available Kinematic viscosity No Data Available

### **SECTION 10: Stability and reactivity**

#### 10.1 Reactivity

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On heating: release of toxic/corrosive/combustible gases/vapors (ammonia). On burning: release of toxic and corrosive gases/vapors (nitrous

vapors). Concentrated solution violent to explosive reaction with many compounds e.g.: with (some) halogens compounds, with (strong) oxidizers and with (some) acids.

#### 10.2 Chemical stability

Ammonia solution itself is not flammable, but can form an ignitable ammonia/air-mixture by outgassing.

#### 10.3 Possibility of hazardous reactions

Reacts vigorously with strong oxidizers and acids.

#### 10.4 Conditions to avoid

High temperature. Incompatible materials.

### 10.5 Incompatible materials

May react violently with acids.

#### 10.6 Hazardous decomposition products

Gaseous ammonia.

# **SECTION 11: Toxicological information**

### Information on toxicological effects

### Acute toxicity

LD50 - oral - rat - 350 mg/kg LDLo - inhalation - human - 5000 ppm

#### Skin corrosion/irritation

Skin - Rabbit

Result: Severe irritations Remarks: (29% solution) Dermatitis Necrosis

#### Serious eye damage/irritation

### Respiratory or skin sensitization

LC50 = 9850 mg/m3 (Rat) 1 hLC50 = 13770 mg/m3 (Rat) 1 h

#### Germ cell mutagenicity

Mutagenic for bacteria and/or yeast.

#### Carcinogenicity

No data available

### Reproductive toxicity

No data available

### STOT-single exposure

No data available

### STOT-repeated exposure

No data available

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### **SECTION 12: Ecological information**

### **Toxicity**

24 Hr LC50 rainbow trout: 0.008 mg/L; 96 Hr LC50 fathead minnow: 8.2 mg/L; 48 Hr LC50 bluegill: 0.024 mg/L; 48 Hr EC50 water flea: 0.66 mg/L

### Persistence and degradability

Readily biodegradable in water. Ozonation in water. Biodegradable in the soil.

### Bioaccumulative potential

No data available.

#### Mobility in soil

No data available.

### **SECTION 13: Disposal considerations**

#### Disposal methods

#### **Product disposal**

Waste material must be disposed of in accordance with the national and local regulations.

Leave chemicals in original containers. No mixing with other waste. Handle uncleaned containers like the product itself.

EPA Waste Code: D002

### **SECTION 14: Transport information**

DOT (US)

UN Number: UN2672

Class: 8

Packing Group: III

Proper Shipping Name: Ammonia solution (with 28% Ammonia)

Reportable quantity (RQ): 1,000 lbs

Marine pollutant: Yes Poison inhalation hazard: No

**IMDG** 

UN Number: UN2672

Class: 8

Packing Group: III

Proper Shipping Name: Ammonia solution (with 28% Ammonia)

**IATA** 

UN Number: UN2672

Class: 8

Packing Group: III

Proper Shipping Name: Ammonia solution (with 28% Ammonia)

### **SECTION 15: Regulatory information**

# 15.1 Safety, health and environmental regulations specific for the product in question

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### **Massachusetts Right To Know Components**

Chemical name: Ammonium hydroxide

CAS number: 1336-21-6

### **New Jersey Right To Know Components**

Common name: AMMONIUM HYDROXIDE

CAS number: 1336-21-6

#### Pennsylvania Right To Know Components

Chemical name: Ammonium hydroxide

CAS number: 1336-21-6

### Canadian Domestic Substances List (DSL)

Chemical name: Ammonium hydroxide ((NH4)(OH))

CAS: 1336-21-6

#### **CERCLA RQ Hazardous Substances**

Ammonium Hydroxide: 1,000 lbs.

### **EPCRA Section 313 Toxic chemicals**

Ammonium Hydroxide

#### CAA 112(r) TO Regulated Chemicals for Accidental Release Prevention

Ammonia (conc 20% or greater): 20,000 lbs.

#### **HMIS Rating**

Ammonium Hydroxide 28%	
HEALTH	3
FLAMMABILITY	1
PHYSICAL HAZARD	0
PERSONAL PROTECTION	

### **NFPA Rating**



## **SECTION 16: Other information**

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